

**PCT**WORLD INTELLECTUAL PROPERTY ORGANIZATION  
International Bureau

A29

## INTERNATIONAL APPLICATION PUBLISHED UNDER THE PATENT COOPERATION TREATY (PCT)

|  |           |  |
|--|-----------|--|
| <b>(51) International Patent Classification <sup>5</sup> :</b><br><b>C08K 3/34, C08L 77/00, 23/02</b>  | <b>A1</b> | <b>(11) International Publication Number:</b> <b>WO 93/04117</b><br><b>(43) International Publication Date:</b> 4 March 1993 (04.03.93)  |
| <b>(21) International Application Number:</b> PCT/US92/06732<br><b>(22) International Filing Date:</b> 12 August 1992 (12.08.92)<br><br><b>(30) Priority data:</b><br>744,035 12 August 1991 (12.08.91) US<br>798,489 26 November 1991 (26.11.91) US<br><br><b>(71) Applicant:</b> ALLIED-SIGNAL INC. [US/US]; 101 Columbia Road, P.O. Box 2245, Morristown, NJ 07962-2245 (US).<br><br><b>(72) Inventors:</b> MAXFIELD, MacRae ; 1226 Cambridge Avenue, Plainfield, NJ 07062 (US). SHACKLETTE, Lawrence, W. ; 11 Alden Place, Maplewood, NJ 07040 (US). BAUGHMAN, Ray, H. ; 14 Glacier Drive, Morris Plains, NJ 07950 (US). CHRISTIANI, Brian, R. ; 139 Indiana Street, Maplewood, NJ 07040 (US).   |           | <b>(74) Agent:</b> ROONEY, Gerard, P.; Allied-Signal, Inc., Law Department (C.A. McNally), 101 Columbia Road, P.O. Box 2245, Morristown, NJ 07962-2245 (US).<br><br><b>(81) Designated States:</b> CA, JP, European patent (AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LU, MC, NL, SE).<br><br><b>Published</b><br><i>With international search report.</i> |
| <b>(54) Title:</b> MELT PROCESS FORMATION OF POLYMER NANOCOMPOSITE OF EXFOLIATED LAYERED MATERIAL<br><br><b>(57) Abstract</b><br><br>This invention relates to a process of forming a polymeric nanocomposite comprising a continuous polymeric phase formed from a melt processible polymer having a melt processing temperature equal to or greater than about 220 °C and platelet particles having an average thickness equal to or less than about 50 Å, and a maximum thickness of about 100 Å having a secondary or primary ammonium cationic complex, a quaternary phosphonium cationic complex bonded to surface of said particles, the composite material formed by said process and an article formed from the composite material. |           |  |

**BEST AVAILABLE COPY**

**FOR THE PURPOSES OF INFORMATION ONLY**

Codes used to identify States party to the PCT on the front pages of pamphlets publishing international applications under the PCT.

|    |                          |    |                                       |    |                          |
|----|--------------------------|----|---------------------------------------|----|--------------------------|
| AT | Austria                  | FI | Finland                               | MN | Mongolia                 |
| AU | Australia                | FR | France                                | MR | Mauritania               |
| BB | Barbados                 | GA | Gabon                                 | MW | Malawi                   |
| BE | Belgium                  | GB | United Kingdom                        | NL | Netherlands              |
| BF | Burkina Faso             | GN | Guinea                                | NO | Norway                   |
| BG | Bulgaria                 | GR | Greece                                | NZ | New Zealand              |
| BJ | Benin                    | HU | Hungary                               | PL | Poland                   |
| BR | Brazil                   | IE | Ireland                               | PT | Portugal                 |
| CA | Canada                   | IT | Italy                                 | RO | Romania                  |
| CF | Central African Republic | JP | Japan                                 | RU | Russian Federation       |
| CG | Congo                    | KP | Democratic People's Republic of Korea | SD | Sudan                    |
| CH | Switzerland              | KR | Republic of Korea                     | SE | Sweden                   |
| CJ | Côte d'Ivoire            | LI | Liechtenstein                         | SK | Slovak Republic          |
| CM | Cameroon                 | LK | Sri Lanka                             | SN | Senegal                  |
| CS | Czechoslovakia           | LU | Luxembourg                            | SU | Soviet Union             |
| CZ | Czech Republic           | MC | Monaco                                | TD | Chad                     |
| DE | Germany                  | MG | Madagascar                            | TG | Togo                     |
| DK | Denmark                  | MI | Mali                                  | UA | Ukraine                  |
| ES | Spain                    |    |                                       | US | United States of America |

MELT PROCESS FORMATION OF POLYMER NANOCOMPOSITE  
OF EXFOLIATED LAYERED MATERIAL

5    1.   Field of the Invention

          This invention relates to a process for forming a composite material comprising a polymer matrix having dispersed therein platelet particles, derived from swellable intercalated layered materials, and to  
10   composite material formed by the process of this invention. More particularly, this invention relates to such a process where the intercalated layered material has layers which are compatible with the polymer of the matrix, such that during the process the  
15   polymer-compatible layers of the intercalated material dissociate one from the other to form platelet fillers.

          2.   Prior Art

          A specific category of polymer nanocomposites has  
20   been described as a composite material comprising a polymer matrix containing a polyamide having uniformly dispersed therein layers of silicate. Such nanocomposites are described in US Pat No. 4,739,007 and No. 4,810,734, Deutsches Pat. 3808623 A1, Japanese  
25   Patent J 02 208358 A, and technical publications by the patentees (J. Inclusion Phenomena 5, (1987), 473-483; Clay Minerals, 23, (1988), 27; Polym. Preprints, 32, (April 1991), 65-66; and Polym. Prints, 28, (August 1987), 447-448.

30

SUMMARY OF THE INVENTION

          This invention relates to a process for forming a polymeric composite which comprises platelet particles dispersed in a polymeric matrix which comprises the  
35   steps of:

          (a) forming a "flowable mixture" comprising a melt-processible polymer and a swellable and polymer-

compatible intercalated layered material having layers that are compatible with said polymer which are compatibilized with one or more "effective swelling/compatibilizing agents" selected from the

5 group consisting of cations of the formula:

$^+\text{NH}_3\text{R}_1$ ,  $^+\text{NH}_2\text{R}_2\text{R}_3$  and  $^+\text{PR}_4\text{R}_5\text{R}_6\text{R}_7$  wherein:

$\text{R}_1$  is an organic radical having at least about 12 aliphatic carbon atoms;

10  $\text{R}_2$  and  $\text{R}_3$  are the same or different and are organic radicals having at least about 5 carbon atoms; and

$\text{R}_4$ ,  $\text{R}_5$ ,  $\text{R}_6$  and  $\text{R}_7$  are the same or different and are organic radicals including at least one which has at least about 8 aliphatic carbon atoms; and

(b) subjecting said mixture to a shear having a  
15 shear rate which is sufficient to dissociate all or a portion of said layers one from the other to form platelet particles having an average thickness equal to or less than about 50 Å, and preferably having a maximum thickness of about 100 Å, and to uniformly  
20 disperse said platelet particles in said polymer to form said polymer composite.

As used herein, "melt processing temperature" means the polymer has a melt viscosity of less than about 5000 Pascal se at a shear rate of 100 sec<sup>-1</sup> as measured  
25 by capillary rheometry when the polymer is in the melt.

The process is directed toward preparation of polymer composites wherein the dispersed phase comprises platelet particles having two flat opposite faces, the thickness of which particles is the distance  
30 between these faces. The thickness is relatively small compared to the size of the flat opposite faces. Dispersing such finely divided platelet particles imparts a very large area of contact between polymer and particles for a given volume of particles in the  
35 composite and a high degree of homogeneity of the composite with respect to the particular effect of the dispersed particles. Platelet particles of high

strength and modulus, dispersed at sub-micron size (nanoscale), impart greater mechanical reinforcement to the polymer matrix than do comparable loadings of conventional reinforcing fillers of micron size.

- 5 Nanoscale barrier layers impart lower permeability to polymers than do comparable loadings of conventional barrier fillers.

The process of this invention exhibits several advantages over monomer blending and polymerizing processes of the type described in U.S. Patent Nos. 10 4,810,734 and 4,739,007, also referred to herein as in-reactor processes. These advantages include utility for a broader range of matrix polymers; utility for a wider range of composites having the same matrix 15 polymer due to a larger selection of swelling/compatibilizing agents, each having a distinct bonding interaction with both the polymer and the platelet particle; and greater control over the molecular weight distribution of the matrix polymer. 20 For example, virtually any polymer material that can be made to flow can be compounded with nanoscale particles derived from intercalated layered materials which exfoliate during mixing in accordance with the process of this invention. In contrast, the monomer blending 25 and polymerizing processes of the prior art are restricted to polymers whose monomers are compatible with layered materials and can be polymerized effectively in the presence of the layered material. In the process of the present invention, the 30 compounding entails no special conditions specific to selected polymer molecular weight distributions. On the other hand, in-reactor processes of the prior art require special polymerization conditions for each selected molecular weight distribution due to the 35 effect of a dispersed phase on reaction mixture viscosity and polymerization kinetics. Virtually any loading of intercalated layered material is possible in

the process of this invention , whereas in-reactor  
compounding of the prior art processes may be  
practically limited to loadings that permit  
polymerization to proceed to a selected molecular  
5 weight.

Another advantage results from the fact that the  
swelling/compatibilizing agents used in this invention  
are secondary and primary ammonium and quaternary  
phosphonium cation complexes containing specific number  
10 of aliphatic carbon atoms. Use of these  
swelling/compatibilizing agents provides for several  
advantages over other process where the  
swelling/compatibilizing agents are tertiary and  
quaternary ammonium cation complexes, and secondary and  
15 primary ammonium and quaternary phosphonium cation  
complexes containing less than the required number of  
aliphatic carbon atoms. For example, the specific  
swelling/compatibilizing agents of this invention cover  
the layers of the layered materials to render their  
20 surfaces more organophilic than those compatibilized by  
tertiary and quaternary ammonium cation complexes and  
those compatibilized with secondary and primary  
ammonium cation complexes and quaternary phosphonium  
cation complexes having less than the required number  
25 of aliphatic carbon atoms. The specific  
swelling/compatibilizing agents of this invention  
facilitate exfoliation of the layered material into  
platelet particles in the polymer melt employing  
reduced shear mixing resulting in less decomposition of  
30 the polymer and reduction in molecular weight. The  
specific swelling/compatibilizing agents of this  
invention are more heat stable than other cationic  
swelling/compatibilizing agents such as tertiary and  
quaternary ammonium cation complexes and secondary and  
35 primary ammonium and quaternary phosphonium cationic  
complexes having less than the specified number of  
aliphatic carbon atoms. The result is that the

swelling/compatibilizing agents of this invention do not substantially decompose at melt processing temperatures equal to or greater than about 220°C into lower molecular weight materials which degrade polymers in the matrix or which are hazardous, either as vapors evolved during melt processing, or in articles such as films in food and drug packing.

As a further advantage of the present process, unreacted monomer can be removed prior to forming the nanocomposite. This facilitates monomer removal, for example by solvent extraction, since dispersed particles can interfere with this process.

Another aspect of this invention relates to a composite material comprising a polymeric matrix which comprises a melt processible polymer having uniformly dispersed therein platelet particles having an average thickness equal to or less than about 50 Å and preferably having a maximum thickness equal to or less than about 100 Å, said platelets having surfaces which are compatibilized with one or more "effective swelling/compatibilizing agents" selected from the group consisting of cations of the formula:

$^+NH_3R_1$ ,  $^+NH_2R_2R_3$  and  $^+PR_4R_5R_6R_7$  wherein:

$R_1$  is an organic radical having at least about 12 aliphatic carbon atoms;

$R_2$  and  $R_3$  are the same or different and are organic radicals having at least about 5 carbon atoms; and

$R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are the same or different and are organic radicals having at least one which has at least about 8 aliphatic carbon atoms.

The polymeric compositions of this invention exhibit one or more advantages over prior art composites as for example those described in U.S. Patent Nos. 4,739,007; 2,531,396 and 4,410,734; Deutsches Pat. 3,808,623 A1; Japanese Patent No. 02208358A and EPA 0,398,551; 0,358,415; 0,352,042 and 0,398,551. For example, the composite of this

invention exhibits improved properties such as improved tensile yield strength, tensile modulus and/or ultimate elongation. In addition, they exhibit superior ability to fix acidic dyes.

5

DESCRIPTION OF THE PREFERRED EMBODIMENTS  
OF THE INVENTION

The first step of this invention comprises forming a "flowable mixture" comprising one or more polymers in  
10 a "polymer melt" and at least one "swellable and polymer-compatible intercalated layered material" which comprises polymer-compatible layers that are compatible with said polymers. As used herein, a "flowable mixture" is a mixture which is capable of flowing at  
15 the submicron scale so that the layered materials may exfoliate into platelet particles comprising individual or a small multiple of layers, which may in turn disperse within the polymer mixture. As used herein, a "polymer melt" is a melt processible polymer or mixture  
20 of polymers which has been heated to a temperature sufficiently high to produce a viscosity low enough for submicron scale mixing to occur. Temperatures used in the first step are not critical and can be varied widely as desired provided that the polymer employed is  
25 in the state of a polymer melt. In the preferred embodiments of the invention, process temperature should be at least as high as the melting point of the particular polymer employed, and below the degradation temperature of the polymer. In the more preferred  
30 embodiments of this invention, where the polymer is a thermoplastic polymer, the process temperature is such that the polymer will remain in the polymer melt during the conduct of the process. In the case of a crystalline thermoplastic polymers, the temperature is  
35 above the polymer's melting temperature. For example, a typical nylon 6 having a melting point of about 225°C can be melted in an extruder at any temperature



equal to or greater than about 225°C, as for example between about 225°C and about 325°C. For nylon 6, a temperature of preferably from about 250°C to about 260°C is normally employed. In the cases of amorphous thermoplastics and vulcanizable rubbers, it is a temperature at which the viscosity is sufficiently low that processing of the polymer can be performed by conventional means.

The manner in which the flowable mixture is formed is not critical and conventional methods can be employed. For example, the flowable mixture can be prepared through use of conventional polymer and additive blending means, in which the polymer is heated to a temperature sufficient to form a polymer melt and combined with the desired amount of the intercalated layered material in a granulated or powdered form in a suitable mixer, as for example an extruder, a Banbury Mixer, a Brabender mixer, a continuous mixer and the like. The polymer melt containing nano-dispersed delaminated layered material may also be formed by reactive extrusion in which the layered material is initially dispersed as aggregates or at the nanoscale in a liquid or solid monomer and this monomer is subsequently polymerized in an extruder or the like. Such monomer or other reactive solid or liquid dispersion can be injected into a polymer melt containing one or more polymers in an extruder or other mixing device. The injected liquid may result in new polymer or in chain extension or grafting to the polymer initially in the melt. Alternatively, the polymer may be granulated and dry mixed with the intercalated layered material, and thereafter, the composition may be heated in a mixer until the polymer is melted forming the flowable mixture. As described above, the flowable mixture is, in the second step, subjected to a shear in a mixer sufficient to form the

dispersed nanocomposite structure of platelet particles in the polymer melt, and it is thereafter cooled.

Best results are generally obtained when the flowable mixture includes as little water as possible in order to avoid hydrolytic cleavage of the polymer and/or generation of void defects. Some polymers such as polyolefins can tolerate more than about 0.25% water by weight during melt processing, while others (such as condensation polymers as for example polyamides and polyesters) may be degraded if processed with water content greater than about 0.1% by weight. Consequently, in the most preferred embodiments, both the polymer and the intercalated layered material are rigorously dried and contain substantially no water. Ideally, the intercalated layered material is free of water that can be desorbed at temperatures up to the processing temperature. However, good results can be obtained when the intercalated layered material contains as much as about 2% by weight water and comprises less than about 5% by weight of the flowable mixture. For compounding with condensation polymers, the intercalated layered material preferably contains less than about 1% by weight, more preferably less than about 0.5% by weight water and most preferably less than about 0.25% by weight water.

As a first essential ingredient, the flowable mixture includes a "swellable and polymer-compatible intercalated material". As used herein, a "swellable and polymer-compatible intercalated layered material" is a swellable layered material intercalated by a neutral or ionic intercalant or intercalants which act to weaken the interlayer cohesive energy by swelling the interlayer distances and which function to increase the compatibility and bonding of the layers with the polymer matrix by having attractive interactions with both the layers and the polymer.

Swellable layered materials are materials comprising planar layers arrayed in a coherent, coplanar structure, where the bonding within the layers, is stronger than the bonding between the layers such that the materials exhibit increased interlayer spacing in their intercalation compounds. The effective swelling/compatibilizing agents may be introduced into the interlayer spaces by either insertion, in the case of neutral molecules, or ion exchange, in the case of ions. The effective swelling/compatibilizing agents may be introduced in the form of a solid, liquid, gas, or solute. The effective swelling/compatibilizing agents may be introduced into the spaces between every layer, nearly every layer, or a large fraction of the layers of the layered material such that the resulting platelet particles comprise less than about 10 layers in thickness. The platelet particles are preferably less than about 8 layers in thickness, more preferably less than about 5 layers in thickness, and most preferably, about 1 or 2 layers in thickness.

Any swellable layered material having the above referenced characteristics may be used in the practice of this invention. Useful swellable layered materials include phyllosilicates. Illustrative of such materials are smectite clay minerals such as montmorillonite, nontronite, beidellite, volkonskoite, hectorite, saponite, sauconite, magadiite, and kenyaite; vermiculite; and the like. Other useful layered materials include illite minerals such as ledikite and admixtures of illites with the clay minerals named above. Other useful layered materials, particularly useful with anionic polymers, are the layered double hydroxides, such as  $\text{Mg}_6\text{Al}_{3.4}(\text{OH})_{18.8}(\text{CO}_3)_{1.7} \cdot \text{H}_2\text{O}$  (see W.T. Reichle, J. Catal., 94 (12985) 547), which have positively charged layers and exchangeable anions in the interlayer spaces. Other layered

materials having little or no charge on the layers may be useful in this invention provided they can be intercalated with swelling agents which expand their interlayer spacing. Such materials include chlorides  
5 such as  $\text{ReCl}_3$  and  $\text{FeOCl}$ , chalcogenides such as  $\text{TiS}_2$ ,  $\text{MoS}_2$ , and  $\text{MoS}_3$ , cyanides such as  $\text{Ni}(\text{CN})_2$ , and oxides such as  $\text{H}_2\text{Si}_2\text{O}_5$ ,  $\text{V}_6\text{O}_{13}$ ,  $\text{HTiNbO}_5$ ,  $\text{Cr}_{0.5}\text{V}_{0.5}\text{S}_2$ ,  $\text{W}_{0.2}\text{V}_{2.8}\text{O}_7$ ,  $\text{Cr}_3\text{O}_8$ ,  $\text{MoO}_3(\text{OH})_2$ ,  $\text{VOPO}_4 \cdot 2\text{H}_2\text{O}$ ,  $\text{CaPO}_4\text{CH}_3 \cdot \text{H}_2\text{O}$ ,  $\text{MnHAsO}_4 \cdot \text{H}_2\text{O}$ ,  $\text{Ag}_6\text{Mo}_{10}\text{O}_{33}$ , and the like.

10 Preferred swellable layered materials are those having charges on the layers and exchangeable ions such as sodium cations, quaternary ammonium cations, calcium cations and the like between the layers which can be intercalated by effective swelling/compatibilizing  
15 agents by an ion exchange mechanism. More preferred layered materials are those having negative charges or basic sites on the layers, preferably at least about 20 basic sites per 100 g of material, more preferably at least about 50 basic sites per 100 g of material and  
20 most preferably from about 50 to about 120 basic sites per 100 g of material. Most preferred swellable layered materials are phyllosilicates having a negative charges on the layers ranging from about 0.2 to about  
25 of exchangeable cations in the interlayer spaces. Particularly preferred layered materials are smectite clay minerals such as montmorillonite, nontronite, beidellite, volkonskoite, hectorite, saponite, sauconite, magadiite, and kenyaite, with hectorite and  
30 montmorillonite having from about 20 basic sites to about 150 basic sites per 100 g material being the layered material of choice.

Swellable layered materials, such as the preferred smectite clay materials generally require treatment by  
35 one or more intercalants to provide the required interlayer swelling and/or polymer compatibility. The resulting interlayer spacing is critical to the

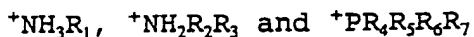
performance of the intercalated layered material in the practice of this invention and interlayer spacing must be sufficiently large to allow for the desired exfoliation of the layers during the process. As used  
5 herein the "inter-layer spacing" refers to the distance between the faces of the layers as they are assembled in the intercalated material before any delamination (or exfoliation) takes place. The preferred clay materials generally include interlayer or exchangeable  
10 cations such as  $\text{Na}^+$ ,  $\text{Ca}^{+2}$ ,  $\text{K}^+$ ,  $\text{Mg}^{+2}$  and the like. In this state, these materials do not delaminate in host polymer melts regardless of mixing, because their interlayer spacings are usually equal to or less than about 4 Å consequently the interlayer cohesive energy  
15 is relatively strong. Moreover, the metal cations do not aid compatibility between layers and the polymer melt. In the preferred embodiments, these layered materials are intercalated by swelling agents of sufficient size to increase interlayer spacing to the  
20 desired extent. In general, the interlayer spacing should be at least about 4 Å, as determined by x-ray diffraction, in order to facilitate delamination of the layered material at the nanoscale. In the preferred embodiments of the invention, the interlayer spacing is  
25 at least about 6 Å and more preferred interlayer spacings are at least about 8 Å. Most preferred interlayer spacings are equal to or greater than about 10 Å. In the embodiments of choice interlayer spacings are at least about 15 Å.

30 In order to further facilitate delamination of layered materials into platelet particles and prevent reaggregation of the particles, these layers are intercalated by the effective swelling/compatibilizing agents of this invention. These agents consist of a  
35 portion which bonds to the surface of the layers and another portion which bonds or interacts favorably with the polymer in the matrix. The agent of this invention

remains bonded to the surface of the layers during and after melt processing as a distinct interphase that is different from the bulk of the polymer matrix. Such agents preferably include a moiety or moieties which interact with the surface of the layers displacing, totally or in part, the original metal ions and which bonds to the surface of the layers; and includes a moiety or moieties whose cohesive energies are sufficiently similar to that of the polymer that the surface of the platelets is made more compatible with the polymer, thereby enhancing the homogeneity of the dispersion in the polymeric matrix. As used herein "compatible" refers to the extent to which the polymer matrix and the surface coating on the platelet particles (the compatibilizing agent) have a favorable interaction which promotes the intermingling of the matrix polymer and the surface layer in the interphase region. Compatibility derives from one or more of the following criteria: similar cohesive energy densities for the polymer and the derivatized platelets, similar or complimentary capacities for dispersive, polar, or hydrogen bonding interactions, or other specific interactions, such as acid/base or Lewis-acid/Lewis-base interactions. Compatibilization will lead to an improved dispersion of the platelet particles in the matrix and an improved percentage of delaminated platelets with a thickness of less than 50 Å.

The nature of the swelling/compatibilizing agent, swelling agent and/or compatibilizing agent will vary widely depending on the particular polymer and the particular layered material. The effective swelling/compatibilizing agent of this invention is selected from the group consisting of cations of the formulas:

35



wherein:

$R_1$  is an organic radical having at least about 12 aliphatic carbon atoms;

$R_2$  and  $R_3$  are the same or different and are organic radicals having at least about 5 carbon atoms; and

5  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are the same or different and are organic radicals of which at least one has about 8 aliphatic carbon atoms.

Such ammonium and phosphonium radicals are well known in the art and can be derived from the  
10 corresponding amines and phosphines using conventional processors.

Illustrative of such suitable  $R_1$ ,  $R_2$ ,  $R_3$ ,  $R_4$ ,  $R_5$ ,  $R_6$  AND  $R_7$  groups are suitable organic radical as for example alkyl, such as methyl, ethyl, octyl, nonyl, 15 tert-butyl, neopentyl, isopropyl, sec-butyl, dodecyl and the like; alkenyl such as 1-propenyl, 1-butenyl, 1-pentenyl, 1-hexenyl, 1-heptenyl, 1-octenyl and the like; alkoxy such as propoxy, butoxy, methoxy, isopropoxy, pentoxy, nonoxy, ethoxy, octoxy, and the 20 like; cycloalkenyl such as cyclohexenyl, cyclopentenyl and the like; alkanoylalkyl such as butanoyloctadecyl, pentanoylnonadecyl, octanoyl pentadecyl, ethanoylundecyl, propanoyl hexadecyl and the like; amino; aryl such as phenyl, naphthyl and the like; 25 alkylaryl such as nonylphenyl, octylphenyl tert-butylphenyl and like; alkylaminoalkyl, such as methylamino octadecyl, ethylamino pentadecyl, butylamino nonadecyl and the like; dialkylaminoalkyl, such as dimethylamino octadecyl, methylethylamino 30 nonadecyl and the like; arylaminoalkyl such as phenylamino octadecyl, p-methylphenylamino nonadecyl and the like; diarylaminoalkyl, such as diphenylamino pentadecyl, p-nitrophenyl-p'-methylphenylamino octadecyl and the like; alkylarylaminomalkyl, such as 35 2-phenyl-4-methylamino pentadecyl and the like; alkylsulfinyl, alkylsulfonyl, alkylthio, arylthio, arylsulfinyl, and arylsulfonyl such as butylthio

octadecyl, neopentylthio pentadecyl, methylsulfinyl  
 nonadecyl, benzylsulfinyl pentadecyl, phenylsulfinyl  
 octadecyl, propylthiooctadecyl, octylthio pentadecyl,  
 nonylsulfonyl nonadecyl, octylsulfonyl hexadecyl,  
 5 methylthio nonadecyl, isopropylthio octadecyl,  
 phenylsulfonyl pentadecyl, methylsulfonyl nonadecyl,  
 nonylthio pentadecyl, phenylthio octadecyl, ethylthio  
 nonadecyl, benzylthio undecyl, phenethylthio  
 pentadecyl, sec-butylthio octadecyl, naphthylthio  
 10 undecyl and the like; alkoxycarbonylalkyl such as  
 methoxycarbonyl, ethoxycarbonyl, butoxycarbonyl and the  
 like; cycloalkyl such as cyclohexyl, cyclopentyl,  
 cyclo-octyl, cycloheptyl and the like; alkoxyalkyl such  
 as methoxy-methyl, ethoxymethyl, butoxymethyl,  
 15 propoxyethyl, pentoxybutyl and the like; aryloxyalkyl  
 and aryloxyaryl such as phenoxyphenyl, phenoxymethyl  
 and the like; aryloxyalkyl such as phenoxydecyl,  
 phenoxyoctyl and the like; arylalkyl such as benzyl,  
 phenethyl, 8-phenyloctyl, 10-phenyldecyl and the like;  
 20 alkylaryl such as 3-decylphenyl, 4-octylphenyl, 4-  
 nonylphenyl and the like; aliphatic acid functions such  
 as  $-C_3H_7COOH$ ,  $-C_5H_{11}COOH$ ,  $-C_7H_{15}COOH$ ,  $-C_7H_{13}COOH$ ,  $-$   
 $C_9H_{19}COOH$ ,  $-C_{11}H_{23}COOH$ ,  $-C_{13}H_{27}COOH$ ,  $-C_{15}H_{31}COOH$  and  $-C_{17}H_{35}COOH$   
 and a moiety of the formula:

25



wherein  $R_8$  is alkyl, cycloalkyl, aryl,  $R_q$  is hydrogen,  
 alkyl, or aryl,  $q$  is an integer equal to or greater  
 30 than 1 and  $Z$  is  $-O-$  or  $-NR_{10}-$ , where  $R_{10}$  is hydrogen,  
 alkyl, aryl or alkylsilane and the like.

Useful swelling/compatibilizing agents may be non-  
 reactive with the polymer matrix but having certain  
 attractive interactions with the polymer matrix such as  
 35 entanglements, hydrogen bonding, or other specific  
 interactions such as acid/base or Lewis acid/Lewis base  
 interactions and the like. Other useful



swelling/compatibilizing agents are reactive with a polymeric component in the polymer to form covalent bonds between matrix the swelling/compatibilizing agent and the polymeric component, or, in the case of

5 secondary ammonium cations and quaternary phosphonium cations may include both reactive and non-reactive moieties. Useful reactive swelling/compatibilizing agents includes one or more substituents selected from the group consisting of organic radicals which are

10 compatible with the polymer forming the composite and which has substituents which are reactive with the polymer such as nucleophilic or electrophilic moieties which are capable of electrophilic or nucleophilic displacement reactions coupling and ring opening

15 reactions and the like as for example amino, carboxy, carboxylic acid, oxide, alkenyl, acetylene, acylhalide, acyloxy, hydroxy, isocyanato, ureido, halo, epoxy, epichlorohydrin, sulfuryl halide, mercapto, ester, and the like.

20 Swelling/compatibilizing agents containing reactive substituents are well known in the art and are selected for particular classes of matrix polymers based on the reactivity and specifically of the reactive moiety. For example polyamides are acid

25 terminated and/or amine terminated and polyesters are acid terminated or hydroxy terminated. Thereof, reactive function groups which react with acid, amine or hydroxy functions to form covalent bonds can be conveniently used with polyesters and polyamides.

30 Specific reactive functions such as  $-NH_2$ ,  $-N=C=O$ ,  $-CONH_2$ ,  $-OH$ ,  $OM^+$  (where M is a metal cation),  $-C(O)X$  (where X is Cl, Br, or I) and the like in the reactive silane compound react with reactive functionalities contained in polyesters and polyamide. Similarly,

35 swelling, compatibilizing agents of this invention containing functions such as  $-NH_2$ ,  $CH_2-X$  (where X is Cl, Br or I),  $-CH_2=CH_2$ ,  $-SH$ ,  $SM^+$  (where  $M^+$  is a metal cation

such as  $\text{Na}^+$ ,  $\text{Li}^+$  and  $\text{K}^+$ ) and  $\text{S}_4\text{H}$  can be conveniently reacted with polyolefins and halogenated polyolefins such as polyethylene, polypropylene, poly(chlorotrifluoroethylene) or polyolefin elastomers to form covalent bonds between the polyolefins and the reactive silanes. Likewise, polyvinyls such as poly(vinyl chloride), poly(co-ethylene vinyl alcohol) and the like can be reacted with swelling/compatibilizing agents of this invention containing  $-\text{NH}_2$ ,  $-\text{CH}_2-\text{S}$  (wherein X is Cl, Br and I)  $-\text{OH}$ ,  $-\text{OM}^+$  (where M is a metal cation such as  $\text{Li}^+$ ,  $\text{Na}^+$  and  $\text{K}^+$ ), and the like to form covalent bonds between the polymers and the swelling/compatibilizing agent of this invention.

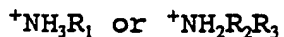
Layered material derivatized with primary ammonium cations having less than 12 carbons are less preferred in most polymer mets except with exhaustive shear mixing. More preferred are those having 16 or more carbons. Most preferred are those having 18 or more carbons and those having more than one ammonium cation group. Illustrative of primary ammonium cations are protonated primary amines such as octadecylamine, arginine and diamine terminated polyethylene telomere (molecular weight of about 2600, ACX 614 obtained from Allied-Signal).

Organic radicals preferred secondary ammonium cations are those having four or more carbons. Most preferred are those having five or more carbons and those having more than one ammonium group. Illustrative of these secondary ammonium cations are protonated amines such as dipentyl amine, bis(ethylhexyl)amine, piperidine, piperazine, hexamethyleneimine, and partially protonated polyethyleneimine. Secondary ammonium cations may comprise that one or more polymer-reactive moieties. Illustrative of preferred secondary ammonium cations

that are polymer-reactive are protonated 11-(hexylamino)undecanoic acid and bis(8-hexanol) amine.

Organic radicals preferred in phosphonium cations are disclosed in U.S. Patent No. 4,136,103. They preferably include at least one having 8 or more carbons. More preferably, they include at least 2 groups of 8 or more carbons or at least one group of 16 or more carbons. Most preferably they include at least one group of 18 or more carbons. Illustrative of these phosphonium cations are octyltributylphosphonium, hexadecyltributylphosphonium and vinylbenzyltriethylphosphonium. One or more of the moieties may be polymer-reactive. Illustrative of preferred phosphonium cations having reactive moieties are bis-(dimethylaminopropyl) dodecyl isobutylphosphonium and bis-(hydroxypropyl) octadecylisabutyl phosphonium).

The swelling/compatibilizing agents are preferably selected from the group consisting of primary and secondary ammonium cationic complexes of the formula:



wherein:

$\text{R}_1$  is an aliphatic radical having at least about 15 aliphatic carbon atoms, said radical optionally including one or more heteroatoms, carbonyl functions or a combination thereof; and

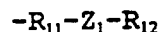
$\text{R}_2$  and  $\text{R}_3$  are the same or different and are aliphatic radicals having at least about four aliphatic carbon atoms, or  $\text{R}_2$  and  $\text{R}_3$  together may form a divalent aliphatic chain having at least about four aliphatic carbon atoms forming an alicyclic structure, said radical or chain optionally including one or more heteroatoms, carbonyl function or a combination thereof.

In the preferred embodiments of the invention  $\text{R}_1$  is an aliphatic radical having at least about 18 carbon atoms, said aliphatic radical optionally includes one

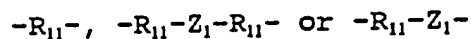
or heteroatoms, carbonyl functions or a combination thereof; and

$R_2$  and  $R_3$  are the same or different and are aliphatic radicals having at least five aliphatic carbon atoms or  $R_2$  and  $R_3$  together may form a divalent aliphatic chain having at least about five aliphatic carbon atoms, said aliphatic radicals and chain may optional include one or more heteroatoms, carbonyl functions or a combination thereof.

In the particularly preferred embodiments of the invention  $R_1$ ,  $R_2$  and  $R_3$  are the same or different and are cyclozlkyl, cycloalkenyl, cycloalkynyl, alkyl, alkenyl or alkynyl or a moiety of the formula:



or  $R_2$  and  $R_3$  together may form a divalent moiety of the formula:



completing an atcyclic ring, wherein:

$-R_{11}-$  is the same or different at each occurrence and is divalent, alkylene, cycloalkylene, cycloalkenylene, alkenylene or alkynylene;

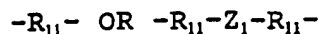
$-R_{12}$  is alkyl, alkylaryl, alkoxyaryl, alkenyl, alkynyl, aryl, cycloalkyl, or cycloalkenyl;

$-Z_1-$  is  $-O-$ ,  $-NR_{13}-$ ,  $-^+N(R_{13})_2-$ ,  $-S-$ ,  $-S(O)_2-$ ,  $-OC(O)-$  or  $-N(R_{13})C(O)-$  and

$R_{13}$  is hydrogen or alkyl having from 1 to about 4 carbon atoms.

In the most preferred embodiments of the invention:

$R_1$ ,  $R_2$  and  $R_3$  are the same or different and are alkyl; or  $R_2$  and  $R_3$  together may form a divalent moiety of the formula:



$-R_{11}-$  is alkylene;

$Z_1$  is  $-O-$ ,  $-NH-$  or  $-^+(R_{13})_2-$ ;

$R_{13}$  is hydrogen or alkyl of from 1 to about 4 carbon atoms.

Layered material may be derivatized by a single swelling/compatibilizing agent of this invention, or by a mixture of such agents, or by a mixture of one or more thermally stable onium cations with one or more other swelling/compatibilizing agents such as one or more organo silanes or quaternary or tertiary ammonium radicals as for example such swelling/compatibilizing agents which have moieties which are compatible with and which are optionally reactive with the polymer forming the matrix. In the preferred embodiments of the invention the moieties are such that the swelling and compatibilizing agents are lipophilic such that the surface tension, at 20°C, or the derivatized particle is preferably less than or equal to about 55 dyne/cm. More preferably, between about 55 and 15 dyne/cm, and preferably between 45 and 20 dyne/cm, as determined by measuring the contact angles made by sessile drops of liquids on the solid surfaces.

Illustrative of such optional onium compounds are oxonium compounds of the formula:



wherein  $X^+$  is a tertiary or quaternary ammonium radical and  $R_{14}$  is an radical as for example substituted or unsubstituted alkyl, cycloalkenyl, cycloalkyl, aryl, or alkylaryl, either unsubstituted or substituted with amino, alkylamino, dialkylamino, nitro, azido, alkenyl, alkoxy, cycloalkyl, cycloalkenyl, alkanoyl, alkylthio, alkyl, aryloxy, arylalkylamino, alkylamino, arylamino, dialkylamino, diarylamino, aryl, alkylsulfinyl, aryloxy, alkylsulfinyl, alkylsulfonyl, arylthio, arylsulfinyl, alkoxycarbonyl, arylsulfonyl, alkylsilane, and a moiety of the formula:



wherein  $R_{16}$  is alkyl, cycloalkyl, or aryl,  $R_{15}$  is hydrogen, alkyl, or aryl,  $q$  is an integer equal to or greater than 1 and  $Z$  is  $-O-$  or  $-NR_{17}-$ , where  $R_{17}$  is hydrogen, alkyl, aryl or alkylsilane. Illustrative of

another optional class of swelling/compatibilizing agents useful in the practice of this invention are silane coupling agents such as those of the formula:



where  $\text{R}_{18}$  and  $\text{R}_{19}$  is the same or different at each occurrence and are alkyl, alkoxy or oxysilane such as trialkoxysilane compounds as for example octadecyltrimethoxysilane, gamma-aminopropyl-  
10 triethoxysilane, gamma-aminopropyltrimethoxysilane, gamma-aminopropylphenyldimethoxysilane, gamma-glycidoxypropyl tripropoxysilane, 3,3-epoxycyclohexylethyl trimethoxysilane, gamma-propionamido trithoxysilane, N-trimethoxysilylpropyl-  
15 N(beta-aminoethyl) amine, trimethoxysilylundecylamine, trimethoxy silyl-2-chloromethylphenylethane, trimethoxysilyl-ethylphenylsulfonylazide, N-trimethoxysilylpropyl-N,N,N-trimethylammonium chloride, N-(trimethoxysilylpropyl)-N-methyl-N,N-diallylammonium  
20 chloride, trimethoxysilylpropylcinnamate, 3-mercaptopropyl trimethoxysilane, 3-isocyanatopropyl-triethoxysilane, and the like; and  $\text{R}^8$  is selected from the group consisting of organic radicals which are compatible with the polymer forming the composite.

25 The amount of swelling agent/compatibilizing agent and swelling/compatibilizing agents intercalated into swellable layered materials useful in this invention may vary substantially provided that the amount is effective to swell and, preferably to compatibilize the  
30 layers of the intercalated layered material to the extent required to provide the desired substantially uniform dispersion. In the preferred embodiments of the invention where swelling/compatibilizing agents are employed, amounts of agents employed will preferably  
35 range from about 10 mmole/100 g of layered material to about 1000 mmole/100 g of layered material. More preferred amounts are from about 20 mmole/100 g to

about 200 mmole/100 g. In the case of the preferred smectite clay minerals, the more preferred amounts are from about 80 mmole/100 g to about 120 mmole/100 g of layered material.

- 5        Swellable and polymer-compatible intercalated layered material can be formed by any method. Preferably such materials are formed by intercalation of suitable agents or agents in the interlayer spaces of the swellable layered material by any suitable  
10 method. The swelling/compatibilizing agents are introduced into the interlayer spaces of the swellable layered material by any suitable method as, for example, by either insertion of neutral molecules or by  
15 ion exchange with ionic molecules, using conventional procedures. Insertion of neutral molecules may be performed by exposing finely divided layered material to intercalants in the form of a gas, neat liquid, finely divided solid, or solute in a solvent which, preferably swells the layered material. Insertion is  
20 generally aided by exposure of the mixture of intercalant and layered material to heat, ultrasonic cavitation, or microwaves. Ion exchange by ionic molecules may be performed by suspending the layered material in a relatively volatile liquid which is  
25 capable of both exfoliating and dispersing the layers of the intercalated layered material and dissolving a salt of the ionic intercalant as well as the resulting salt of the ion displaced from the layered material (e.g.,  $\text{Na}^+$ ,  $\text{Mg}^{+2}$ ,  $\text{Ca}^{+2}$ ), adding the salt of the ionic  
30 intercalant, and removing the layered material (now complexed with the new intercalant) from the liquid (now containing the dissolved salt of the displaced ion). For example, swellable layered minerals such as montmorillonite and hectorite (having primarily  $\text{Na}^+$   
35 cations in the interlayer spaces) intercalate water to the point that the layers are exfoliated and dispersed uniformly in water. Dispersion in water is generally

aided by mixing with relatively high shear. A suitable swelling/compatibilization agent such as the hydrochloride salt of dipentylamine is then added in the desired amount after which the layers complexed with the ammonium cation are separated from the dispersion, washed of residual NaCl, and dried. In the preferred embodiments of the invention, the swellable layered material is intercalated by ion exchange. For example, a suspension of a montorillonite or a saponite in water, may be heated to about 80°C and stirred using a high speed homogenizer mixer, in a concentration low enough to yield a low viscosity dispersion from which non-dispersible particles can be separated by sedimentation (mineral concentration of about 2% by weight, or 5% to 15% with addition of a peptizing agent such as sodium hexametaphosphate). The dispersion is combined with a solution of a suitable swelling/compatibilizing agent such as an ammonium salt (as, for example the hydrochlorides of octadecylamine, 11-aminoundecanoic acid, dioctylamine, piperidine, dipentylamine, and the like such that the mole ratio of ammonium salt to exchangeable ions in the mineral is between 0.5 and 5. The amine-complexed layers may be separated from the solution by some suitable method such as filtration or centrifugation, followed by rinsing in fresh water, rough drying, and ball milling to about 100 mesh powder. The powder may be rigorously dried at 100°C to 140°C in vacuum for 8 to 24 h in the presence of a drying agent such as phosphorous pentoxide, to provide the desired swellable/polymer compatible intercalated layered material.

Intercalated layered materials intercalated with a mixture of the agents of this invention and silanes may be formed by treating a swellable and polymer-compatible intercalated layered material already intercalated with an agent of this invention with a silane coupling agent in a swelling liquid, such as



dioxane, glyme, diglyme, dimethylsulfoxide, methylethylketone, and the like, or by treating an aqueous suspension of a layered material with water-soluble silane coupling agents such as trialkoxysilanes followed by interaction with an agent of this invention. In the preferred embodiments, silane intercalated swellable/polymer compatible intercalated layered material is formed as follows: Layered materials intercalated with the swelling/compatibilizing agent of this invention, preferably prepared as described above are suspended and swollen in a swelling organic liquid and treated with a trialkoxysilane. For example, montmorillonite intercalated with octadecylammonium cation, at about 80 mmole of ammonium cation/100 g mineral, is combined with dioxane to form a 5% by weight suspension which is heated to 60°C and combined with a dioxane solution of aminoethylaminopropyl trimethoxysilane, such that the ratio of silane to mineral is about 20 mmole/100 g. The silane displaces the ammonium cation quantitatively to form a mixed intercalated layered material having about 60 mmole of ammonium cation and 20 mmole of silane per 100 g of mineral layers.

In the preferred embodiments of this invention, swellable and polymer-compatible intercalated layered compounds include montmorillonite (Gelwhite HNF, Southern Clay Products) complexed with octadecylammonium cation (100 mmole/100 g mineral), montmorillonite complexed (Volclay, American Colloids Company) with dipentylammonium cation (100 mmole/100 g), synthetic hectorite (Laponite S, Laporte Industries) complexed with dioctylammonium cation (80 mmole/100 g), commercially available organo clay (Claytone APA<sup>®</sup>, Southern Clay Products), montmorillonite complexed with octadecylammonium cation (about 80 mmole/g) and derivatized with

aminoethylaminopropyltrimethoxysilane (20 mmole/100 g), and the like.

The amount of intercalated layered material included in the mixture may vary widely depending on the intended use of the composite. The amount of intercalated layered material included in the mixture is generally at least about 0.001% by weight of the composite, more preferably from about 0.01 to about 20% by weight of the composite and most preferably from about 0.1 to about 10% by weight of the composite. The amount of material employed in any particular situation will depend to a significant extent on the intended use. For example, relatively, larger amounts of platelet particles (exclusive of intercalant since the intercalant content in the layered material may vary), i.e. from about 15% to about 30% by wgt. of the mixture, are used in applications where articles are formed by stamping. Substantially enhanced barrier properties and heat resistance (deflection temperature under load, DTUL) are imparted by platelet particle concentrations greater than about 2.5%. Similarly, substantially enhanced strength is imparted by platelet particle concentrations greater than about 1.5%. When it is desired to preserve such properties as toughness (impact resistance) and elongation which are generally adversely affected by high loadings of any filler material including the nano-scale layered materials of this invention, it is preferred that the silicate loading be less than about 0.5%. Particle concentration within the range 0.05 to 0.5% significantly enhance modulus, dimensional stability, and wet strength (the latter in the case of polyamides). Concentrations below 0.5% can be employed to increase melt viscosity (useful in film extrusion and in fiber melt spinning) or they may be employed in selected polymers to stabilize a particular crystalline phase (useful in the case of nylon 6 to stabilize the

gamma phase) or limit spherulite size which reduces haze and increases optical clarity. In general, the amount of material employed is less than about 60% by weight of the mixture. The amount of material employed is preferably from about 0.01% to about 20 % by weight of the mixture, more preferably from about 0.05% to about 10% by weight of the mixture, and most preferably from about 0.05% to about 8% by weight.

The second essential ingredient of the flowable mixture is a melt processible polymer. Polymers for use in the process of this invention may vary widely, the only requirement is that they are melt processible. As used herein, a "polymer" is a substance composed of ten or more recurring monomeric units which may be the same or different which is melt processible, preferably at a temperature equal to or greater than about 220°C, preferably equal to or greater than about 230°C, more preferably from about 230° to about 320°C, and most preferably from about 240° to about 290°. In the preferred embodiments of the invention, the polymer includes at least twenty recurring monomeric units. The upper limit to the number of recurring monomeric units is not critical, provided that the melt index of the polymer under use conditions is such that the polymer forms a flowable mixture. More preferably, the polymer includes at least about 30 recurring monomeric units. In the most preferred embodiments of this invention the number of recurring units is such that the polymer has a melt index of from about 0.01 to about 12 grams per 10 minutes at the processing temperature.

Useful polymers are thermoplastic polymers or mixtures thereof, and vulcanizable and thermoplastic rubbers. Thermoplastic resins for use in the practice of this invention may vary widely. Illustrative of useful thermoplastic resins are polylactones such as poly(pivalolactone), poly(caprolactone) and the like;

polyurethanes derived from reaction of diisocyanates such as 1,5-naphthalene diisocyanate, p-phenylene diisocyanate, m-phenylene diisocyanate, 2,4-toluene diisocyanate, 4,4'-diphenylmethane diisocyanate, and the like and linear long-chain diols such as poly(tetramethylene adipate), poly(ethylene adipate), poly(1,4-butylene adipate), poly(ethylene succinate), polyether diols and the like; polycarbonates such as poly[methane bis(4-phenyl) carbonate], poly[1,1-ether bis(4-phenyl) carbonate], poly[diphenylmethane bis(4-phenyl) carbonate], poly[1,1-cyclohexane bis(4-phenyl) carbonate] and the like; polysulfones; polyether ether ketones; polyamides such as poly(4-amino butyric acid), poly(hexamethylene adipamide), poly(6-aminohexanoic acid), poly(m-xylylene adipamide), poly(metaphenylene isophthalamide) (Nomex), poly(p-phenylene terephthalamide) (Kevlar), and the like; polyesters such as poly(ethylene-1,5-naphthalate), poly(1,4-cyclohexane dimethylene terephthalate), poly(ethylene oxybenzoate) (A-Tell), poly(para-hydroxy benzoate) (Ekonol), polyethylene terephthalate, poly(butylene terephthalate) and the like; poly(arylene oxides) such as poly(2,6-dimethyl-1,4-phenylene oxide), poly(2,6-diphenyl-1,4-phenylene oxide) and the like; poly(arylene sulfides) such as poly(phenylene sulfide) and the like; polyetherimides; vinyl polymers and their copolymers such as polyvinyl alcohol, polyvinyl chloride, polyvinylidene chloride, ethylene-vinyl acetate copolymers, and the like; polyacrylics, polyacrylate and their copolymers such as polyethyl acrylate, poly(n-butyl acrylate), poly(ethyl methacrylate), polyacrylamide, polyacrylonitrile, methacrylate-styrene copolymers, and the like; polyolefins such as low density poly(ethylene), poly(propylene), poly(4-methyl-1-pentene), poly(styrene), and the like; ionomers; poly(epichlorohydrins); poly(urethanes) such as the

polymerization product of diols such as glycerin, trimethylol-propane, 1,2,6-hexanetriol, and the like with a polyisocyanate such as 2,4-tolylene diisocyanate, 2,6-tolylene diisocyanate, 4,4'-diphenylmethane diisocyanate, 1,6-hexamethylene diisocyanate, 4,4'-dicyclohexylmethane diisocyanate and the like; and polysulfones such as the reaction product of the sodium salt of 2,2-bis(4-hydroxyphenyl) propane and 4,4'-dichlorodiphenyl sulfone; furan resins such as poly(furan); cellulose ester plastics such as cellulose acetate, cellulose acetate butyrate, cellulose propionate and the like; silicones such as poly(dimethyl siloxane), poly(dimethyl siloxane), poly(dimethyl siloxane co-phenylmethyl siloxane), and the like; protein plastics; and blends of two or more of the foregoing.

Vulcanizable and thermoplastic rubbers useful in the practice of this invention may also vary widely. Illustrative of such rubbers are brominated butyl rubber, chlorinated butyl rubber, polyurethane elastomers, fluoroelastomers, polyester elastomers, butadiene/acrylonitrile elastomers, silicone elastomers, poly(butadiene), poly(isobutylene), ethylene-propylene copolymers, ethylene-propylene-diene terpolymers, sulfonated ethylene-propylene-diene terpolymers, poly(chloroprene), poly(2,3-dimethylbutadiene), poly(butadiene-pentadiene), chlorosulphonated poly(ethylenes), poly(sulfide) elastomers, block copolymers, made up of segments of glassy or crystalline blocks such as poly(styrene), poly(vinyl-toluene), poly(t-butyl styrene), polyester and the like and the elastomeric blocks such as poly(butadiene), poly(isoprene), ethylene-propylene copolymers, ethylene-butylene copolymers, polyether ester and the like as for example the copolymers in poly(styrene)-poly(butadiene)-poly(styrene) block

copolymer manufactured by Shell Chemical Company under the trade name of Kraton®

In the preferred embodiments of the invention, the polymers of choice are polymers and copolymers of olefins, polyesters, polyamides and blends thereof. In the particularly preferred embodiments of the invention, polymers and copolymers of ethylene and propylene (preferably polyethylene) and poly(propylene) and more preferably polyethylene, polyamides (preferably nylon 6 and nylon 66 and more preferably nylon 6), and blends thereof are used.

The particular preferred polyamide useful in the preferred embodiments of the invention has a melt index from about 0.01 to about 10 grams/10 minutes, and preferably from 0.5 to 10 grams/10 minutes as measured by ASTM Test No. D-1238 at a load of 1000 grams at 235°C. Amongst these particularly preferred embodiments, most preferred are those embodiments in which the polyamide is nylon 6 or nylon 66, with nylon 6 being the polyamide of choice. The particularly preferred homopolymers or copolymers of ethylene and propylene have a melt index of from about 0.01 to about 1.0 grams per 10 minutes, preferably from about 0.05 to about 0.8 grams per 10 minutes as measured according to ASTM Test No. D1238 at a load of 2160 grams at 190°C. Amongst these particularly preferred embodiments, most preferred are those in which the polymer is polyethylene or polypropylene, with polyethylene being the polymer of choice.

The mixture may include various optional components which are additives commonly employed with polymers. Such optional components include nucleating agents, fillers, plasticizers, impact modifiers, chain extenders, plasticizers, colorants, mold release lubricants, antistatic agents, pigments, fire retardants, and the like. These optional components and appropriate amounts are well known to those of

skill in the art, accordingly, only the preferred optional components will be described herein in detail.

In the second step of the process of this invention, the flowable mixture is subjected to a shear  
5 having an "effective shear rate". As used herein, an "effective shear rate" is a shear rate [as shear rate is defined in Ferdinand Rodriguez, "Principles of Polymers Systems", McGraw-Hill Book Company, New York (1982)] which is effective to delaminate at least about  
10 90% by weight of the intercalated material to form platelet particles described above, and provide a composition comprising a polymeric matrix having platelet particles substantially homogeneously dispersed therein. In the preferred embodiments of the  
15 invention, the shear rate is greater than about  $10 \text{ sec}^{-1}$ . In these preferred embodiments of the invention, the upper limit for the shear rate is not critical provided that the shear rate is not so high as to physically degrade the polymer. In the particularly  
20 preferred embodiments of the invention, the shear rate is from greater than about  $10 \text{ sec}^{-1}$  to about  $20,000 \text{ sec}^{-1}$ , and in the most preferred embodiments of the invention the shear rate is from about  $100 \text{ sec}^{-1}$  to about  $10,000 \text{ sec}^{-1}$ .

25 Any method which can be used to apply a shear to a flowable mixture or any polymer melt can be used. The shearing action can be provided by any appropriate method, as for example by mechanical means, by thermal shock, by pressure alteration, or by ultrasonics.  
30 Methods useful in shearing melts are known in the art, and will not be described in great detail. In particularly useful procedures, the flowable polymer mixture is sheared by mechanical methods in which portions of the melt are caused to flow past other  
35 portions of the mixture by use of mechanical means such as stirrers, Banbury® type mixers, Brabender® type mixers, long continuous mixers, and extrudes. Another

procedure employs thermal shock in which shearing is achieved by alternatively raising or lowering the temperature of the mixture causing thermal expansions and resulting in internal stresses which cause the shear. In still other procedures, shear is achieved by sudden pressure changes in pressure alteration methods; by ultrasonic techniques in which cavitation or resonant vibrations which cause portions of the mixture to vibrate or to be excited at different phases and thus subjected to shear. These methods of shearing flowable polymer mixtures and polymer melts are merely representative of useful methods, and any method known in the art for shearing flowable polymer mixtures and polymer melts may be used.

In the preferred embodiments of the invention, mechanical shearing methods are employed such as by extrusion, injection molding machines, Banbury® type mixers, Brabender® type mixers and the like. In the more preferred embodiments of the invention, shearing is achieved by introducing the polymer melt at one end of the extruder (single or double screw) and receiving the sheared polymer at the other end of the extruder. The temperature of the polymer melt, the length of the extruder, residence time of the melt in the extruder and the design of the extruder (single screw, twin screw, number of flights per unit length, channel depth, flight clearance, mixing zone etc.) are several variables which control the amount of shear to be applied.

Upon subjecting a flowable mixture of said swellable intercalated layered material and said polymer melt to shear mixing, at least about 80% by weight, preferably at least about 85% by weight, more preferably at least about 90% by weight and most preferably at least about 95% by weight of the layers of the material delaminate to form platelet particles substantially homogeneously dispersed in the polymer



matrix. As used herein, "platelet particles" are particles having two relatively flat opposite faces wherein the thickness of which is the distance between the faces, which is relatively small compared to the size of the faces. As formed by this process, the platelet particles dispersed in matrix polymers have the thickness of the individual layers, or small multiples less than about 10, preferably less than about 5 and more preferably less than about 3 of the layers, and still more preferably 1 or 2 layers. In the preferred embodiments of this invention, intercalation of every interlayer space is complete so that all or substantially all individual layers delaminate one from the other to form separate platelet particles. In cases where intercalation is incomplete between some layers, those layers will not delaminate in the polymer melt, and will form platelet particles comprising those layers in a coplanar aggregate. These latter platelet particles still constitute nanoscale and nanodispersed fillers and provide enhanced properties over and above those provided by conventional micro-scale fillers, as long as they are less than about 10 layers thick and preferably less than 5 layers thick. The other dimensions of the platelet particles may vary greatly, but in the case of particles derived from clay minerals, the particle faces are roughly round or oblong having average diameters between about 10,000 Å and about 50 Å, such that the aspect ratio length/thickness ranges from about 1000 to about 1. For the purposes of the present invention, the average diameter is defined as the diameter of a circle having an area equal to the surface area of one broad surface face of the platelet shaped particle. The average diameter is determined from particle surface area as measured with a Leitz Texture Analyzer System in a fully computerized and automated mode. In the preferred embodiments of the

invention the average thickness of the platelet particles is equal to or less than about 20 Å and the average diameter is between 5,000 Å and 100 Å. Most preferably the average thickness is about 10 Å. The  
5 most preferred average diameter depends on both the desired property of the nanocomposite and the ease of complete intercalation and delamination to form the nanocomposite structure. High aspect ratios, and therefore large average diameters, are generally  
10 preferred for reinforcement and barrier properties, while layered materials having smaller platelets are preferred for their ease of delamination. Thus, for purposes of the nanocomposite properties, the most preferred average diameter is greater than about 150 Å,  
15 and, for purposes of delamination, it is less than about 3000 Å.

The average interparticle spacing between delaminated layers may vary widely after shearings, depending on the concentration of layered material. In  
20 general the higher the concentration of layered material in the polymer matrix particle the smaller the interparticle spacing; and conversely, the lower the concentration of layered material, the larger the interparticle spacing. In general, interparticle  
25 spacing is equal to or greater than 15Å. The interparticle spacing is preferably equal to or greater than about 20Å more preferably equal to or greater than about 30Å and most preferably equal to or greater than about 50Å.

30 As used herein "uniformly dispersed" is defined as a degree of dispersion of the platelet shaped particles having a standard deviation in platelet particle density, down to a sampling volume of  $10^{-15} \text{ m}^3$ , which is preferably less than about 50% of the mean, more  
35 preferably less than about 30% of the mean, and most preferably less than about 20% of the mean as

etermined from estimates based on transmission electron microscopy.

The process of this invention is preferably carried out in the absence of air, as for example in the presence of an inert gas, such as, argon, neon, nitrogen or the like. The process can be carried out in a batchwise or discontinuous fashion, as for example, carrying out the process in a sealed container. Alternatively, the process can be carried out in a continuous fashion in a single processing zone, as for example by use of an extruder, from which air is largely excluded, or in a plurality of such reaction zones in series or parallel.

The nanocomposites of this invention exhibit useful properties which are superior to those predicted by U.S. Patent Nos. 4,739,007 and 5,810,734. Various useful performance indices can be devised in order to jointly assess different performance aspects using a single number. Depending upon the nature of the devised index, comparisons can be made either between samples containing the same loading of layered material or the index can be more broadly applied to polymers having different loadings of clay. For example, the effect of adding nanoscale particulate fillers dispersed platelet particles to a polymer typically increases tensile modulus and ultimate tensile strength while decreasing ultimate elongation. In the case where combinations of high modulus ( $Y$ ) and high ultimate elongation ( $\Delta L/L$ ) are required, a useful performance index for comparing samples with similar particulate loading is  $Y(\Delta L/L)$ . Likewise, when tensile strength ( $S$ ) combined with high ultimate elongation is sought, a useful performance index for comparing samples with similar loading levels is  $S(\Delta L/L)$ . The preferred embodiments of the present invention provide a  $Y(\Delta L/L)$  of above about 660 MPa (preferably equal to or greater than about 800 MPa more preferably equal to

or greater than about 1000 MPa and most preferably equal to or greater than about 1200 MPa) for a loading of about 2% of nanodispersed layered material such as montmorillonite in a melt processible polymer such as nylon 6. In contrast, the prior art patents on in-reactor nanocomposite blends (U.S. 4,739,007 and U.S. 4,810,734) provide a maximum  $Y(\Delta L/L)$  of about 660 MPa. Also, preferred embodiments of the present invention provide a  $S(\Delta L/L)$  of above about 20 MPa (preferably equal to or greater than about 25 MPa and more preferably equal to or greater than about 30 MPa) for a loading of about 2% intercalated layered material such as montmorillonite, while the maximum value for this figure of merit obtained in the above mentioned in-reactor process patents is about 20 MPa at 2% loading. While we do not wish to be bound by any theory, it is believed that the unique properties of the nanocomposites of this invention result from the use of swelling/compatibilizing agents which do not react to become part of the main chain of the matrix polymer (as do the swelling agents preferred in the prior art U.S. 4,739,007 and U.S. 4,810,734) when used in the in-reactor processes. Instead, the preferred swelling/compatibilizing agents of this invention, which remain bonded to the particle surface, interact with the matrix polymer in one or more of the following ways: (1) covalent bond formation to form branches on the main chains of the matrix polymer, (2) hydrogen, ion-dipole, and dipole-dipole bonding with portions of the matrix polymer, and (3) Van der Waals attraction and entanglement with the matrix polymer. Nanocomposites having particularly attractive combinations of modulus, tensile strength and ultimate elongation are formed using swelling/compatibilizing agents, under category (3) above, which are bonded to the particle surface, and whose polymer-interacting moiety projects away from the particle surface and is

only weakly bonded to the matrix polymer. This function is best performed by secondary ammonium compounds of the formula  $^+NH_2R_1$  primary ammonium compounds of the formula:  $^+NH_2R_2R_3$  and quaternary phosphonium compounds of the formula  $^+PR_4R_5R_6R_7$  of the type having a lipophilic moiety such that the surface tension, at 20°C, of the derivatized particle is less than about 55 dyne/cm, preferably between about 55 and 15 dyne/cm, and more preferably between about 45 and 20 dyne/cm, as determined by measuring the contact angles made by sessile drops of liquids on solid surfaces. Illustrative of these preferred secondary ammonium compounds, primary ammonium compounds and quaternary phosphonium of the formula:

$^+NH_2R_1$ ,  $NH_2R_2R_3$  or  $^+PR_4R_5R_6R_7$   
wherein  $R_1$ , at least one of  $R_2$  and  $R_3$  or at least one of  $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  is alkyl, alkenyl, cycloalkyl, alkoxyalkyl, alkynyl, phenylalkyl, alkoxyphenyl, alkenylphenyl, phenylalkenyl, phenylalkynyl, alkynylphenyl or the like, preferably alkyl which include sufficient number of aliphatic carbon atoms such that the radical on the particle has a surface tension at 20°C of less than about 55 dyne/cm.

The nanocomposite compositions according to the invention are thermoplastic and, in some cases, vulcanizable materials from which molded articles of manufacture having valuable properties can be produced by conventional shaping processes, such as melt spinning, casting, vacuum molding, sheet molding, injection molding and extruding. Examples of such molded articles are components for technical equipment, apparatus castings, household equipment, sports equipment, bottles, containers, components for the electrical and electronics industries, car components, circuits, fibers, semi-finished products which can be shaped by machining and the like. The use of the materials for coating articles by means of powder

coating processes is also possible, as is their use as hot-melt adhesives. The molding compositions according to the invention are outstandingly suitable for specific applications of all types since their spectrum of properties can be modified in the desired direction in manifold ways. Such molded products of this invention will derive one or more advantages over products molded with polymers having no nanodispersed platelet particles including increased modulus, stiffness, wet strength, dimensional stability, and heat deflection temperature, and decreased moisture absorption, flammability, permeability, and molding cycle time.

The molding compositions according to the invention are outstandingly suitable for the production of sheets and panels having valuable properties. Such sheets and panels may be shaped by conventional processes such as vacuum processing or by hot pressing to form useful objects. The sheets and panels according to the invention are also suitable as coating materials for other materials comprising, for example, wood, glass, ceramic, metal or other plastics, and outstanding strengths can be achieved using conventional adhesion promoters, for example, those based on vinyl resins. The sheets and panels can also be laminated with other plastic films and this is preferably effected by co-extrusion, the sheets being bonded in the molten state. The surfaces of the sheets and panels, including those in the embossed form, can be improved or finished by conventional methods, for example by lacquering or by the application of protective films.

The compositions of this invention are especially useful for fabrication of extruded films and film laminates, as for example, films for use in food packaging. Such films can be fabricated using conventional film extrusion techniques. The films are preferably from about 10 to about 100 microns, more

preferably from about 20 to about 100 microns and most preferably from about 25 to about 75 microns in thickness. In the film, the major plane of the platelet fillers is substantially parallel to the major plane of the film. The extent of parallelism of particles and film can be determined by X-ray analysis. X-ray analysis is a useful way to described the crystallinity and orientation of polymer crystals and the orientation of platelet particles. A convenient method of X-ray analysis is that described in Hernans, P.H. and Weidinger A., Makromol Chemie, Vol. 44, pp. 24-36 (1961), hereby incorporated by reference.

For the purpose of the present invention Op, the platelet orientation factor, is an indication of the platelet particle orientation in the film. The Op was determined by making azimuthal scans from densitometer tracings of the X-ray photographs which were obtained by exposing the edge of the film to the incident X-rays. The angle is the angle between the reference direction, the normal to the film, and the normal to the plane of interest, the major plane of the platelet. The Op values were calculated as the average cosine square ( $\langle \cos^2 \rangle$ ) for the normal to the flat faces of the platelet particles. An Op of 1.0 indicates that the faces of the platelets are completely parallel to the plane of the film. An Op of 0.0 indicates that the faces of the platelets are perpendicular to the plane of the film. The Op of the platelets in the film of the present invention is preferably from about 0.70 to about 1.0, more preferably from about 0.90 to about 1.0 and most preferably from about 0.95 to about 1.0. Such preferred orientation of platelet particles results in enhanced barrier properties and increased tare strength.

The homogeneously distributed platelet particles and polymer are formed into a film by suitable film-forming methods. Typically, the composition is melted

and forced through a film forming die. The die can be a flat die or a circular die. A typical flat die is a hanger shaped die, and a typical circular die is a tubular film die.

5       The film of the nanocomposite of the present invention may go through steps to cause the platelets to be further oriented so the major planes through the platelets are substantially parallel to the major plane through the film. A method to do this is to biaxially stretch the film. For example, the film is stretched 10 in the axial or machine direction by tension rollers pulling the film as it is extruded from the die. The film is simultaneously stretched in the transverse direction by clamping the edges of the film and drawing 15 them apart. Alternatively, the film is stretched in the transverse direction by using a tubular film die and blowing the film up as it passes from the tubular film die. The films of this invention may exhibit one or more of the following benefits: increased modulus, 20 wet strength, and dimensional stability, and decreased moisture adsorption and permeability to gases such as oxygen and liquids such as water, alcohols and other solvents.

      The following specific examples are presented to 25 more particularly illustrate the invention and are not to be construed as limitations thereon.

#### EXAMPLE 1

      A nylon 6 nanocomposite of a layered material 30 derivatized with a secondary ammonium cation was prepared by compounding, in a melt extrusion process, nylon 6 (Capron 8209F obtained from Allied-Signal) with 4% (w/w) of montmorillite derivatized with dipentyl ammonium cation.

35       The organoclay was prepared by combining an aqueous solution of dipentyl ammonium chloride with a 5% aqueous dispersion of montmorillonite (Gel White HNF,



obtained from Southern Clay Products) at about 80°C with high shear mixing. The stoichiometry of the exchange reaction was 0.125 mole of the dipentyl ammonium chloride per 100g of clay. The organoclay flocculated immediately and, after standing overnight at room temperature, was collected by filtration. The organoclay was washed 4 times with hot water until free of chloride ions, dried in air at 120°C, ground to pass through a 75 micron sieve, and further dried at 120°C (full vacuum) for 18 hours just prior to compounding. Samples of the montmorillonite-dipentyl ammonium cation complex powder were evaluated by thermogravimetric analysis at a heating rate of 10 degree C per minute in inert atmosphere. The apparent decomposition temperature and the extent of decomposition during heating to 300°C are set forth in Table 1.

TABLE 1  
THERMOGRAVIMETRIC ANALYSIS OF  
INTERCALATED MONTMORILLONITES

| Exp. No. | swelling/compatibilizing cation | decomposition onset temperature (C) | weight loss between 100 and 300°C (% of total onium ion content) |
|----------|---------------------------------|-------------------------------------|--|
| 1        | dipentyl ammonium (secondary)   | 275                                 | 5.6  |

a. Claytone APA (montmorillonite complex) obtained from Southern Clay Products.

According to the results summarized in Table 1, the montmorillonite complex of the secondary ammonium cation was more thermally stable than either the tertiary or the quaternary ammonium cation complexes. The montmorillonite-dipentyl ammonium cation complex powder was dry mixed with nylon 6 pellets, and this mixture was extruded using a Leistritz twinscrew extruder equipped with general purpose screws. The

extruder was adjusted as follows: heat zones 1 - 9 at 220-230°C, die 250°C, RPM 250. The pelletized extrudate was molded into test samples. Dry as molded samples were tested for their tensile properties according to the procedures of ASTM D638, and for their heat deflection temperature under a load of 264 psi (DTUL 264 psi) according to the procedures of ASTM D648. The results are set forth in Table 2.

TABLE 2NYLON 6-MONTMORILLONITE NANOCOMPOSITES

| Exp. No. | swelling /compatibilizing agent | weight % montmorillonite <sup>a</sup> | Tensile Modulus, psi (MPa) | Tensile Strength, psi (MPa) | Ultimate Elongation % | DTUL 264 psi, C |
|----------|---------------------------------|---------------------------------------|----------------------------|-----------------------------|-----------------------|-----------------|
| 1        | dipentyl ammonium               | 2.95                                  | 489,300 (3,370)            | 11,600 (80)                 | 44                    | 72              |

a. Determined by Loss on Ignition.

According to the results summarized in Table 2, the montmorillonite complex of the secondary ammonium cation imparted superior elongation along with comparable tensile modulus, tensile yield strength, and heat deflection under load to those exhibited by nylon 6 nanocomposites of montmorillonite complexed with an acidic omega-aminoacid.

COMPARATIVE EXAMPLE 1

A tertiary alkylammonium cation complex of montmorillonite was prepared according to the procedure of Example 1 except that an aqueous solution of dimethyldodecylamine hydrochloride was added to the montmorillonite dispersion. The dried, powdered complex was tested by thermogravimetric analysis, and the results are set forth in Table 3. For comparison purposes, a quaternary alkylammonium cation complex obtained from Southern Clay Products Inc. under the Trademark Claytone APA was also subjected to thermogravimetric analysis. The results for these

analysis together with the thermogravimetric results from Table 1 of Example 1 are set forth in Table 3.

TABLE 3  
THERMOGRAVIMETRIC ANALYSIS OF  
INTERCALATED MONTMORILLONITES

| Exp. No. | swelling/compatibilizing cation       | decomposition onset temperature (C) | weight loss between 100 and 300°C (% of total onium ion content) |
|----------|---------------------------------------|-------------------------------------|--|
| 1        | dipentyl ammonium (secondary)         | 275                                 | 5.6  |
| 2        | dimethyldodecyl ammonium (tertiary)   | 190                                 | 27.8   |
| 3        | quaternary alkylammonium <sup>a</sup> | 220                                 | 38.5   |

a. Claytone APA (montmorillonite complex) obtained from Souther Clay Products.

COMPARATIVE EXAMPLE 2

A nylon 6 nanocomposite of a layered material derivatized with an acidic omega-aminoacid cation was prepared by compounding, in a melt extrusion process, nylon 6 with about 3% (w/w) of montmorillite derivatized with protonated 11-aminoundecanoic acid cation. The montmorillonite complex was prepared, compounded, molded and tested according to the procedure of Example 1 except that an aqueous solution of 11-aminoundecanoic acid hydrochloride was used. The results of the tests are set forth in Table 4. For comparison purposes, nylon 6 was subjected to analysis of ultimate elongation, tensile modulus, tensile strength, and heat deflection under load. These results, together with comparable data for nylon 6 having montmorillnite dipentyl ammonium cation complex are set forth in the following Table 4.

**TABLE 4**  
**NYLON 6-MONTMORILLONITE NANOCOMPOSITES**

| Exp. No. | swelling/com patibilizing agent | weight % montmorillonite <sup>a</sup> | Tensile Modulus, psi (MPa) | Tensile Strength, psi (MPa) | Ultimate Elongation % | DTUL 264 psi, C |
|----------|---------------------------------|---------------------------------------|----------------------------|-----------------------------|-----------------------|-----------------|
| 1        | dipentyl ammonium               | 2.95                                  | 489,300 (3,370)            | 11,600 (80)                 | 44                    | 72              |
| 2        | acidic 11-amino-undecanoic acid | 2.5                                   | 498,000 (3,440)            | 11,600 (80)                 | 12                    | 75              |
| 3        | Nylon 6                         | 0                                     | 350,000 (2,414)            | 9,500 (66)                  | 26                    | 55              |

a. Determined by Loss on Ignition.

According to the results summarized in Table 4, the montmorillonite complex of the secondary ammonium cation imparted superior elongation along with comparable tensile modulus, tensile yield strength, and heat deflection under load to those exhibited by nylon 6 nanocomposites of montmorillonite complexed with an acidic omega-amino acid.

WHAT IS CLAIMED IS:

1. Composite material comprising a polymer matrix which comprises a polymer matrix and dispersed platelet particles having an average thickness of less than 50 Å and a maximum thickness of about 100 Å, and having an onium chemical species bonded to them, said chemical species selected from the group consisting of onium compounds of the formula:



- wherein:

$R_1$  is an organic radical having at least about 12 aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof;

- $R_2$  and  $R_3$  are the same or different at each occurrence and are organic radicals at least one of which having at least about 5 carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof; and

- $R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are the same or different and are organic radicals at least one of which has eight or more aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof, said platelet particles being present in an amount less than about 60% by weight of the composite material.

2. Composite material of claim 1, wherein said platelet particles are derived from phyllosilicates.

3. Composite material of claim 2, wherein said phyllosilicates are smectites clay minerals.

4. Composite material of claim 3, wherein said swelling/compatibilizing agent is selected from the group consisting of primary and secondary ammonium cationic complexes of the formula:

- $^+NH_3R_1$  or  $^+NH_2R_2R_3$

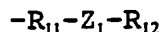
wherein:

$R_1$  is an aliphatic radical having at least about 15 carbon atoms, said aliphatic radical optionally includes one or more heteroatoms, carbonyl functions or a combination thereof; and

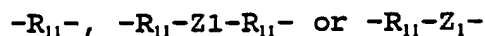
5  $R_2$  and  $R_3$  are the same or different and are aliphatic radicals having at least about five aliphatic carbon atoms or  $R_2$  and  $R_3$  together may form a divalent aliphatic chain having at least about five aliphatic carbon atoms, said aliphatic radicals and chain may  
10 optionally include one or more heteroatoms, carbonyl functions or a combination thereof.

5. Composite material of claim 4, wherein:

$R_1$ ,  $R_2$  and  $R_3$  are the same or different and are cycloalkyl, cycloalkenyl, cycloalkynyl, alkyl, alkenyl,  
15 alkynyl or a moiety of the formula:



or  $R_2$  and  $R_3$  together may form a divalent moiety of the formula:



20 completing an atcyclic ring, wherein:

$-R_{11}-$  is the same or different at each occurrence and is divalent alkylene, cycloalkylene, cycloalkenylene, alkenylene or alkynylene;

$R_{12}$  is alkyl, alkylaryl, alkoxyaryl, alkenyl,  
25 alkynyl, aryl, cycloalkyl, or cycloalkenyl;

$-Z_1-$  is  $-O-$ ,  $-NR_{13}-$ ,  $-N(R_{13})_2-$ ,  $-S-$ ,  $-S(O)_2$ ,  $-OC(O)-$  or  $-N(R_{13})C(O)-$  and

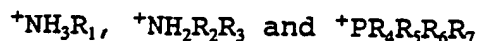
$R_{13}$  is hydrogen or alkyl having from 1 to about 4 carbon atoms.

30 6. Composite material as defined in claim 1, wherein said matrix polymer is selected from the group consisting of polyamides; polyesters; polycarbonates; celluloses; polyolefins; phenolics; poly(urethanes); poly(sulfones); poly(etheretherketones);  
35 poly(esteramides); poly(phenylenesulfides); poly(amideimides); polyacetals; poly(alkylene oxides); poly(phenylene oxides); and poly(imides).

7. Composite material of claim 6, wherein said polymeric matrix comprises a polyamide, a polyester, a polyolefin, or a combination thereof.

8. Composite material of claim 7, wherein said  
5 polymeric matrix comprises poly(ethylene terephthalate), poly(ethylene naphthalate), polyethylene, polypropylene, nylon 6, nylon 6,6, nylon 12 or a combination thereof.

9. An article of manufacture comprising a body,  
10 said body fabricated, totally or in part, from a composite material comprising a polymer matrix which comprises a heat processible polymer and dispersed platelet particles having an average thickness of less than 50 Å and a maximum thickness of about 100 Å, and  
15 having an onium chemical species bonded to them, said chemical species selected from the group consisting of onium compounds of the formula:



wherein:

20  $R_1$  is an organic radical having at least about more 12 aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof;

$R_2$  and  $R_3$  are the same or different at each  
25 occurrence and are organic radicals having at least about 5 carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof; and

$R_4$ ,  $R_5$ ,  $R_6$  and  $R_7$  are the same or different and are  
30 organic radicals at least one of which has 8 or more aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination thereof, said platelet particles being present in an amount less than about 60% by weight of the composite  
35 material.

10. A process for forming a polymeric composite which comprises platelet particles dispersed in a polymeric matrix, said process comprising the steps of:

- (a) forming a flowable mixture comprising a polymer  
5 melt at a temperature equal to or greater than about 220°C and a swellable and polymer-compatible intercalated layered material having an onium species is bonded to the surface of the layers of said material said onium species of the formula:

10  ${}^+\text{NH}_3\text{R}_1$ ,  ${}^+\text{NH}_2\text{R}_2\text{R}_3$  and  ${}^+\text{PR}_4\text{R}_5\text{R}_6\text{R}_7$

wherein:

$\text{R}_1$  is an organic radical at least about 12 aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a combination  
15 thereof;

$\text{R}_2$  and  $\text{R}_3$  are the same or different at each occurrence and are organic radicals at least one of which having at least about 5 carbon atoms and optionally including one or more heteroatoms, carbonyl  
20 functions or a combination thereof; and

$\text{R}_4$ ,  $\text{R}_5$ ,  $\text{R}_6$  and  $\text{R}_7$  are the same or different and are organic radicals at least one of which has eight or more aliphatic carbon atoms and optionally including one or more heteroatoms, carbonyl functions or a  
25 combination thereof, said platelet particles being present in an amount less than about 60% by weight of the composite material; and

- (b) subjecting said mixture to a shear having a shear rate which is sufficient to dissociate all or a  
30 portion of said layers to form platelet particles having an average thickness of less than about 50 Å and to uniformly disperse said platelet particles in said polymer to form said polymer composite wherein said platelet particles are uniformly dispersed in said  
35 matrix.



## INTERNATIONAL SEARCH REPORT

International Application No

PCT/US 92/06732

|   |  |                          |
|---|--|--------------------------|
| <b>I. CLASSIFICATION OF SUBJECT MATTER</b> (If several classification symbols apply, indicate all) *  |  |                          |
| According to International Patent Classification (IPC) or to both National Classification and IPC   |  |                          |
| IPC <sup>5</sup> : C 08 K 3/34, C 08 L 77/00, C 08 L 23/02  |  |                          |
| <b>II. FIELDS SEARCHED</b>  |  |                          |
| Minimum Documentation Searched †  |  |                          |
| Classification System   | Classification Symbols   |                          |
| IPC <sup>5</sup>  | C 08 K, C 08 L 23/00, C 08 L 77/00   |                          |
| Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched *   |  |                          |
| <b>III. DOCUMENTS CONSIDERED TO BE RELEVANT*</b>  |  |                          |
| Category *  | Citation of Document, †† with indication, where appropriate, of the relevant passages †‡   | Relevant to Claim No. ‡‡ |
| P, X  | EP, A1, 0 459 472<br>(KABUSHIKI KAISHA TOYOTA CHUO KENKYUSHO) 04 December 1991<br>(04.12.91),<br>see page 3, line 3 - page 4,<br>line 38; examples; claims.<br>--                  | 1-10                     |
| X   | US, A, 4 889 885<br>(USUKI et al.)<br>26 December 1989 (26.12.89),<br>see column 3, line 11 -<br>column 9, line 41; examples;<br>claims.<br>--                                     | 1-10                     |
| A   | Polymer Preprints, vol. 32,<br>April 1991, American Chem.<br>Soc.<br>KAZUHISA YANO et al. "Synthesis and properties of poly-mide-clay hybrid", pages 65, 66<br>see Totality.<br>-- | 1-10                     |
| <p>* Special categories of cited documents: †‡</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principles or theory underlying the invention</p> <p>"X" document of particular relevance: the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance: the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"Z" document member of the same patent family</p> |  |                          |
| <b>IV. CERTIFICATION</b>  |  |                          |
| Date of the Actual Completion of the International Search   | Date of Mailing of this International Search Report  |                          |
| 03 November 1992  | 25 NOV 1992  |                          |
| International Searching Authority   | Signature of Authorized Officer  |                          |
| EUROPEAN PATENT OFFICE  | TENGLER e.h.   |                          |

| III. DOCUMENTS CONSIDERED TO BE RELEVANT (CONTINUED FROM THE SECOND SHEET) |   |                       |
|--|---|-----------------------|
| Category *   | Citation of Document, " with indication, where appropriate, of the relevant passages  | Relevant to Claim No. |
| A  | <p>EP, A1, 0 351 337<br/> (RHONE-POULENC CHIMIE)<br/> 17 January 1990 (17.01.91),<br/> see examples; claims.<br/> -----</p> | 1-10                  |

## ANHANG

zum internationalen Recherchen-  
bericht über die internationale  
Patentanmeldung Nr.

## ANNEX

to the International Search  
Report to the International Patent  
Application No.

## INNEXE

au rapport de recherche inter-  
national relatif à la demande de brevet  
international n°

PCT/US 92/06732 SAE 63495

In diesem Anhang sind die Mitglieder  
der Patentfamilien der im obenge-  
nannten internationalen Recherchenbericht  
angeführten Patentdokumente angegeben.  
Diese Angaben dienen nur zur Unter-  
richtung und erfolgen ohne Gewähr.

This Annex lists the patent family  
members relating to the patent documents  
cited in the above-mentioned inter-  
national search report. The Office is  
in no way liable for these particulars  
which are given merely for the purpose  
of information.

La présente annexe indique les  
membres de la famille de brevets  
relatifs aux documents de brevets cités  
dans le rapport de recherche inter-  
national visé ci-dessus. Les renseigne-  
ments fournis sont donnés à titre indica-  
tif et n'engagent pas la responsabilité  
de l'Office.

| Im Recherchenbericht<br>angeführtes Patentdokument<br>Patent document cited<br>in search report<br>Document de brevet cité<br>dans le rapport de recherche |         | Datum der<br>Veröffentlichung<br>Publication<br>date<br>Date de<br>publication | Mitglied(er) der<br>Patentfamilie<br>Patent family<br>member(s)<br>Membre(s) de la<br>famille de brevets        | Datum der<br>Veröffentlichung<br>Publication<br>date<br>Date de<br>publication   |
|--|---------|--|---|--|
| EP A1  | 459472  | 04-12-91   | JP A2 4033955   | 05-02-92   |
| US A   | 4889885 | 26-12-89   | JP A2 1198645<br>DE A1 3806548<br>JP A2 63221168<br>JP A2 63215775  | 10-08-89<br>15-09-88<br>14-09-88<br>08-09-88                                     |
| EP A1  | 351337  | 17-01-90   | BR A 8903428<br>CN A 1039251<br>DK A0 3446/89<br>DK A 3446/89<br>FR A1 2634215<br>JP A2 2055745<br>US A 4994518 | 13-02-90<br>31-01-90<br>12-07-89<br>13-01-90<br>19-01-90<br>26-02-90<br>19-02-91 |